# WHAT IS THIS GAS?

Origins	NF <sub>3</sub> is another example of an anthropogenic gas. It is a newly discovered resource in the electronics industry used for computer chips, LCD flat-screens, and solar photovoltaics. NF <sub>3</sub> was introduced as a potential replacement for SF6, CFCs, and HCFCs, but has since shown negative effects on climate change.		
		V1.0 © Net350 LLC 2022	NIT
What is the current PPM of this gas? And the atmospheric trend since 1700 and since 1960?	The current (2020) atmospheric content of NF <sub>3</sub> is approximately 1.7 x $10^{-6}$ PPM. NF <sub>3</sub> measurements began in the mid-late 90s, so there is no measurement for the year 1986. Preindustrial levels of NF <sub>3</sub> are assumably no greater than 0.008 PPT (8e-9 PPM), which is virtually zero.	REDUCTION POTENTIA	
		Can emissions be reduced?	The prac cont Kyot and NF3
Why is this gas important to climate change? How long does it remain in the atmosphere?	$NF_3$ is a new concern regarding climate change. It has a long atmospheric lifetime ranging 500+ years. $NF_3$ has a GWP of 16,000-17,000 over a span of 100 years in terms of heat trapping abilities (Infrared heat gets trapped as it is absorbed, thus contributing to global warming). $NF_3$ is also a harmful substance to human health. Prolonged exposure can result in liver and kidney damage as well as headaches. Because $NF_3$ is used and produced in the electronics industry, it is a growing concern for individuals working closely to it.		for it Unite M. B it cou
		Can emissions be captured at source?	NF₃ gree durir gree perta tech the l pote
What is an appropriate PPM goal to meet 1.5 degree threshold?	$NF_3$ is a relatively recent concern compared to other greenhouse gases such as carbon dioxide and methane. The dangers of $NF_3$ have been acknowledged, but there is no (PPM) number associated with meeting or working towards the 1.5-degree threshold.	Can this gas be removed from the atmosphere or ocean?	Beca conti atmo Ther pote is ce

# 2

# **ROGEN TRIFLUORIDE**

reduction of NF<sub>3</sub> emissions is not widely ticed. NF₃ has been acknowledged as a harmful ributor to climate change. It is not included in the o Protocol, which has been revised many times pushes for globally reduced GHG emissions. is recognized as a greenhouse gas yet demand is increasing. In 2022, the market for NF3 in the ed States alone is estimated to be worth \$1700 v 2029, if there are no implemented regulations, uld be worth up to \$3400 B. was only recently acknowledged as a nhouse gas. Its use was actually encouraged ng the 1990s to reduce the use of other nhouse gases. There is very little information aining to any reduction efforts or capture nologies. NF3 accounts for a small amount of preakdown of greenhouse gases, however, its ncy is what causes the recent concern. ause NF<sub>3</sub> has only been recently considered a ributor to climate change, removal from the osphere and the oceans is widely unknown. e seems to be hypothesized methods and ntial technologies / strategies, however nothing

rtain.



### **TECHNOLOGY REMAINS IN DEVELOPMENTAL STAGES**

What are current and potential removal technologies?

There are no current technologies being used to remove NF<sub>3</sub>. Any and all technologies described are in the experimental stages, therefore, not used on a large scale if used at all. One of these experimental technologies explains using kinetic energy alongside the exposure to ozone over a long period of time to potentially remove NF<sub>3</sub>. Another result from experimental data explained the process of UV photolysis (the decomposition of molecules in light). These removal findings are experimental. While potential removal strategies have been discovered, there is not large-scale technology to continuously expose NF<sub>3</sub> to ozone or allow NF<sub>3</sub> to decompose within the stratosphere.

Which of these technologies are potentially scalable?

NF<sub>3</sub> removal has been studied on a molecular level. Research for the science behind different removal strategies on a molecular level is recently emerging, so there is no technology that has yet to implement these findings. With that being said, removal technologies are essentially non-existent, so there is no scalable technology aside from experiments and studies looking into the science of removal.

## FOR EACH TECHNOLOGY:

What are the uses of the It has only recently been discovered that the most important portion of removal  $(NF_3)$  has to do with removed gas? decomposition when it is exposed to UV light. Because there are no existing removal technologies, there is no known use for removed NF<sub>3</sub>. The science of removal is still being investigated. If the use is sequestration, Because removal research is preliminary, the uses for removed  $NF_3$  is not understood as the potential what is the sequestration time frame? removal sciences are still being studied. What is the net energy requirement to process removal?

The net energy requirement has not been determined. The science of removing NF<sub>3</sub> from certain layers of the atmosphere is still being investigated, but research is beginning to emerge. NF<sub>3</sub> only became a widely used gas in the 1990s to lessen other greenhouse gas emissions. It has been acknowledged as a contributor to climate change, however, it is not recognized in the Kyoto Protocol. Its recent use and barely emerging studies make it difficult to quantify net energy requirements. This quantity is likely in the process of configuration along with potential removal sciences.